## KHYBER PAKHTUNKHWA, PUBLIC SERVICE COMMISSION

## COMPETITIVE EXAMINATION FOR THE POSTS OF PROVINCIAL MANAGEMENT SERVICES (BPS-17).

2013

## **CHEMISTRY PAPER-I**

TIME ATTOWER, ATTOMRS	
TIME ALLOWED: 03 HOURS	MAX:MARKS: 100

	[	
No	te: Attempt any Five questions. All questions carry equal marks.	
Q.	A) Explain Schrodinger wave equation and its graphical representation?	15
	B) Describe postulates of Bohr Theory of Hydrogen atom?	05
Q.	2 A) Explain Standard enthalpies of formation and Hess's Law?	14
	B) What do you know about Born Haber cycle?	06
Q.	A) Explain <b>Debye-Huckel theory</b> , by using this theory define activity coeffice Galvanic cells and Fuel cell?	ients,
	B) Write difference between strong and weak Electrolytes and its effect on cell?	05
Q.	A) What is the role of Carbon in organism and industry?	06
	B) Describe manufacturing of Nitric Acid in laboratory and inclustry?	14
Q	A) Describe method of preparation and purification of Iron ores?	14
	B) Name types of Fertilizer? Explain Phosphorus Fertilizer with its uses?	06
Q.	How the structures of Coordination Compound are explained on the base Molecular Orbital Theory? Explain the strong and weak field Ligands with example of the control of the base of the control of the control of the base of the control of the control of the base of the control	
Q.:	A) What is air pollution and most common diseases related to Air Pollution?	14
	B) Which factors are responsible for Air pollution, how we can overcome its ha effects?	rmful 06
0.8	A) Describe in your words about Fivation of Nitro and (10)	

- words about Fixation of Nitrogen.
  - B) Write down different types of Water hardness. Give names of compounds that are responsible for these types. (10)